

16

Roll No.

24005

**B. Tech. 2nd Semester (Common for all
Branches)**

Examination – May, 2017

ENGINEERING CHEMISTRY

Paper : CH-101-F

Time : Three Hours] [Maximum Marks : 100

Before answering the questions, candidates should ensure that they have been supplied the correct and complete question paper. No complaint in this regard, will be entertained after examination.

Note : Attempt *five* questions in all, selecting *one* question from each Section. Question No. 1 is *compulsory*. All questions carry equal marks.

1. (a) State phase rule. 2 × 10 = 20
(b) What do you understand by autocatalysis ?
(c) Define ppm.
(d) Name any two coagulants.
(e) What is dry corrosion ?
(f) What is meant by rusting of iron?
(g) What are solid lubricants ? Give examples.

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- (h) What is cation polymerization ?
 (i) Give the use of Teflon.
 (j) What is the principle of DTA ?

SECTION - A

2. (a) Draw and explain the phase diagram of Zn-Mg system in detail. 10
 (b) Discuss the phase diagram of any one component system. 10
3. Write short notes on the following : 5 × 4 = 20
 (a) Enzymatic catalysis
 (b) Promoter catalysis
 (c) Poisoners catalysis
 (d) Cooling curve

SECTION - B

4. (a) What is stress corrosion embrittlement in boilers ? How can this be prevented ? 8
 (b) Discuss the scale and sludge formation in boilers and their removal. 12
5. (a) Distinguish between zeolite and Lime-soda softening of hard water. 10
 (b) Give a brief account of various processes used for domestic water treatment. 10

SECTION - C

6. (a) What is cathodic protection? Explain its types for the prevention of corrosion. 10

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- (b) Define electrochemical corrosion. Describe the electrochemical theory of corrosion with reference to rusting of iron. 10

7. (a) What is Lubrication ? Explain the different mechanisms of lubrication. 10
 (b) Describe the following properties of lubricants :
 (i) Flash point and Fire point 5
 (ii) Viscosity index 5

SECTION - D

8. Discuss the following :
 (a) Polymeric composites 10
 (b) Effect of structure on the properties of polymers 10
9. Describe the principle, technique and applications of the following :
 (a) I R Spectroscopy 10
 (b) U V Spectroscopy 10

24005-12900-(P-3)(Q-9)(17) (3)